

Methods of Amination

The present invention relates to methods of amination, and in particular to methods of aminating aromatic groups using 5 transition metal catalysis.

Amine derivatives are exceptionally important pharmaceutical intermediates and active ingredients in many drugs. Aromatic amines form the basis of the modern organic-based 10 photoconductors in xerography (photocopiers and photoconductors) [References 1-4], solar cells and as hole transporting materials in organic and polymeric light emitting devices [References 5-11].

15 Supercritical carbon dioxide and compressed carbon dioxide have emerged as a general environmentally benign solvent for the synthesis of organic molecules [References 12 and 13] and polymers [Reference 14]. It can be particularly beneficial in a variety of palladium-mediated syntheses and cross coupling 20 reactions [References 15-18] and for the integration of synthesis with processing. Particular examples of use in organic electronic materials are described by Ober and DeSimone [References 19-22]. Opportunities for the controlled deposition 25 of organic and polymeric electronic materials have been disclosed [Reference 23]. Deposition from compressed CO<sub>2</sub> will allow the controlled supramolecular ordering of materials owing to the ability to control demixing of samples during deposition from CO<sub>2</sub> solutions.

30 Amination reactions have been historically developed using the Ullmann coupling procedure [References 24 to 27], which involves the copper-mediated coupling of aryl halides and aryl 4-toluenesulfonates. More recently a family of palladium catalysed aromatic amination reactions have been developed in 35 which an aryl halide or aryl tosylate is typically coupled with an amine derivative in the presence of a palladium (0) catalyst,

a suitable bulky organophosphine ligand and a base [Reference 28]. The scope and methodology of such a procedure (the 'Buchwald-Hartwig' amination reaction) has been reviewed by Buchwald and Hartwig [References 29-31] and forms the basis of a 5 wide variety of amine syntheses. The use of these methods for the manufacture of electroactive polymers has been described [Reference 32].

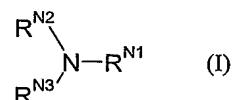
There is an attraction in combining the synthesis of 10 aminederivatives and the subsequent processing in compressed CO<sub>2</sub>. Advantages could include an environmentally friendly manufacturing process plus control of morphology of the final product using anti-solvent techniques (see A. I. Cooper's review [Reference 14]) for pharmaceuticals. In the electroactive 15 organic and polymeric materials arena an advantage of integrated synthesis and processing will lead to architecturally controlled multilayered devices with supramolecular order. A particular example is the use of blended materials to improve organic LED device performance [Reference 33]. Another example of the 20 benefit of an integrated synthesis and processing system is the advantage of polymer deposition where layer separation is required, by virtue of the immiscibility of the deposition solvent with the first layer, or induction of microphase segregation of two materials co-deposited from carbon dioxide 25 whose solubility difference can be exploited to generate organised and phase segregated materials. This feature has specific advantages in organic photovoltaic devices [Reference 34].

30 Although palladium catalysed carbon-carbon bond formation reactions in supercritical CO<sub>2</sub> have been described [Reference 36], prior art in the field would suggest that carrying out the palladium catalysed amination reaction in compressed CO<sub>2</sub> (the Buchwald-Hartwig amination reaction) would fail because it is 35 well known that amines form carbamic acids in the presence of carbon dioxide. In fact, the formation of a carbamic acid has

been used to suppress the reactivity of a free amino substituent in the course of a synthesis in compressed carbon dioxide [Reference 35].

5 The present inventors have now discovered that palladium catalysed amination reactions can be carried in compressed CO<sub>2</sub> by the use of selected N-silylamines.

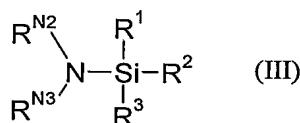
10 Accordingly, the present invention provides a method of synthesising a compound of formula I:



comprising the step of reacting a moiety of formula II:



with a moiety of formula III:



15 in compressed carbon dioxide in the presence of a transition metal catalyst and a base, wherein:

L is a labile leaving group;

R<sup>N1</sup> is optionally substituted C<sub>5-20</sub> aryl;

20 R<sup>N2</sup> is selected from optionally substituted C<sub>5-20</sub> aryl, optionally substituted C<sub>3-20</sub> heterocyclyl, optionally substituted C<sub>3-7</sub> alkyl, and optionally substituted sulfonyl;

R<sup>N3</sup> is selected from H and optionally substituted C<sub>1-7</sub> alkyl, C<sub>3-20</sub> heterocyclyl and C<sub>5-20</sub> aryl; or

25 R<sup>N2</sup> and R<sup>N3</sup> together with the nitrogen atom to which they are attached form optionally substituted nitrogen-containing C<sub>3-20</sub> heterocyclyl or C<sub>5-20</sub> heteroaryl; and

R<sup>1</sup>, R<sup>2</sup> and R<sup>3</sup> are independently selected from optionally substituted C<sub>1-7</sub> alkyl, C<sub>5-20</sub> aryl, C<sub>3-20</sub> heterocyclyl, hydroxy,

30 halo, amino and C<sub>1-7</sub> alkoxy, or two of R<sup>1</sup>, R<sup>2</sup> and R<sup>3</sup>, together with the silicon atom to which they are attached, may form a silicon containing C<sub>5-7</sub> heterocyclyl group (e.g. silacyclobutyl).

5  $R^{N1}$  and  $R^{N2}$  may be linked by a single bond, such that the compound of formula I comprises a nitrogen-containing  $C_{5-7}$ , heterocyclyl or heteroaryl group formed from  $R^{N1}$  and  $R^{N2}$ , and the nitrogen to which they are attached.

It has also been found that these reactions proceed more efficiently than when carried out in an organic solvent, such as toluene.

10

*Compressed carbon dioxide*

The term "compressed carbon dioxide" means herein carbon dioxide which has been compressed under pressure to produce liquid carbon dioxide or supercritical or near supercritical carbon

15 dioxide.

20 A fluid is termed "supercritical" when its temperature exceeds the critical temperature ( $T_c$ ). At this point the two fluid phases, liquid and vapor, become indistinguishable [Reference 37]. The critical temperature of carbon dioxide is 31.1°C and the critical pressure 73.8 bar. Conditions and solvent media required to form supercritical or near supercritical states are described in Reference 12 and References 38 to 45.

25 The reaction is preferably carried out at a pressure between 800psi and 4000psi. More preferably the reaction pressure is greater than, or equal to, 1500 psi. The reaction is also more preferably less than, or equal to, 3500 psi.

30 *Transition metal catalyst*

Suitable transition metal catalysts include complexes of platinum, palladium, iron, nickel, ruthenium and rhodium. Catalyst complexes may include chelating ligands, such as, by way of example only,  $C_{1-7}$  alkyl and  $C_{5-20}$  aryl derivatives of phosphines and bisphosphines, imines, arsines and hybrids thereof, including hybrids of phosphines with amines.

Additionally, heterogeneous catalysts containing forms of these elements are also suitable as catalysts for the present invention. Catalysts containing palladium and copper are preferred, with palladium based catalysts being more preferred.

5

The active form of the transition metal catalyst is not well characterised. Therefore, the term "transition metal catalyst" as used herein refers to any transition metal catalyst and/or catalyst precursor as is introduced into the reaction vessel and 10 which is, if necessary, converted into the active phase, as well as active form, of the catalyst which participates in the reaction.

The palladium catalysts most suitable for use in the present 15 invention are formed from palladium(II) salts and appropriate ligands, preferably phosphine ligands. Such catalysts are known in the art and are described in Reference 12, 36, 38-45. Particularly preferred catalysts include Pd catalysts with one or more phosphine ligands such as  $\text{PPh}_3$ ,  $\text{P}(\text{C}_6\text{H}_{12})_3$ , 2- 20 diphenylphosphinophenol, binap, dppf,  $\text{P}(\text{t-Bu})_2(\text{biphen})$  where biphen represents 2-phenyl-phen-1-yl, where the 2-phenyl group may bear at one or more of the 2', 4' and 6'- positions iso- 25 propyl groups or *N,N*-dimethyl amino groups. Examples of catalysts include, but are not limited to, those derived from  $\text{Pd}(\text{II})$  acetate (especially with  $\text{P}(\text{t-Bu})_2(\text{biphen})$  ligands, where biphen is as defined above),  $\text{Pd}(\text{PPh}_3)_4$ ,  $\text{Pd}(\text{PPh}_3)_2\text{Cl}_2$  and  $\text{Pd}(\text{dppf})\text{Cl}_2$ .

Other preferred transition metal (preferably palladium) 30 catalysts are those based on the *N*-heterocyclic carbenoid ligands described for example by Nolan [Reference 46], and the micro-encapsulated transition metal catalysts disclosed in Reference 36.

The transition metal catalyst is preferably present in the range of 0.001 to 20 mol%, and preferably 1.0 to 2.5 or 5 mol%, with respect to the moiety of formula II.

5      *Base*

Suitable bases for use in the present invention include bases of group 1 metals, carbonate, phosphate or tert-butoxy/phenoxy bases and superbases [References 47 and 48]. Preferred bases are group 1 metal carbonate, phosphate or tert-butoxy/phenoxy 10 bases, such as  $K_2CO_3$ ,  $K_3PO_4$ ,  $Na_2CO_3$ ,  $Cs_2CO_3$ ,  $K(t\text{-}BuO)$ ,  $Na(t\text{-}BuO)$ ,  $K(OPh)$ ,  $Na(OPh)$ , and tetraalkylammonium salts or mixtures thereof.

15      Preferred bases include  $K_2CO_3$ ,  $Na_2CO_3$  and  $Cs_2CO_3$ , of which  $Cs_2CO_3$  is most preferred.

The base is preferably present as 1 to 4 equivalents of the moiety of formula II, and more preferably as 1 to 1.5 or 2 equivalents.

20

*Optional Additive*

The reaction mixture may also contain an optional additive which acts as a fluoride source, to aid the progress of the reaction. Such fluoride sources include, but are not limited to, KF, CsF, 25 tetrabutylammonium fluoride, tris(diethylaminosulfonyl difluorotrimethylsilicate (TASF) and tetrabutylammonium triphenyldifluorosilicate (TBAT), of which KF is most preferred.

30      The optional additive is preferably present as 1 to 2 equivalents of the moiety of formula III, and more preferably as 1 to 1.3 or 1.5 equivalents.

*Labile leaving group*

35      Labile leaving groups suitable for use in the present invention are in particular those known to be amenable to palladium catalysed coupling. Suitable groups include mesylate ( $-\text{OSO}_2\text{CH}_3$ );

-OSO<sub>2</sub>(C<sub>n</sub>F<sub>2n+1</sub>), where n=0-4; -OSO<sub>2</sub>-R<sup>S</sup>, where R<sup>S</sup> is an optionally substituted phenyl group (e.g. 4-Me-Ph, tosylate); -N<sup>+</sup>Me<sub>3</sub>X<sup>-</sup>, where X may be OTf, OTs, I, Br, Cl, OH; I, Br and Cl. More preferred are -OSO<sub>2</sub>(C<sub>n</sub>F<sub>2n+1</sub>) where n=0,1 or 4 (in particular triflate), I, Br and Cl, with Br being the most preferred.

*Amount of compound of formula III*

When the moiety of formula III is not bound to the moiety of formula II, it is preferably present as 1 to 2 equivalents, and 10 is more preferably 1 to 1.3 or 1.5 equivalents, of the compound of formula II.

*Reaction Temperature*

The reaction is preferably carried out at room temperature (i.e. 15 20°C) or higher, more preferably higher than 50°C, but at 200°C or lower. A most preferred temperature range for the reaction is between 60°C and 120°C, with temperatures of about 100°C being particularly preferred.

20 *Substituents*

The phrase "optionally substituted" as used herein, pertains to a parent group which may be unsubstituted or which may be substituted.

25 Unless otherwise specified, the term "substituted" as used herein, pertains to a parent group which bears one or more substituents. The term "substituent" is used herein in the conventional sense and refers to a chemical moiety which is covalently attached to, or if appropriate, fused to, a parent 30 group. A wide variety of substituents are well known, and methods for their formation and introduction into a variety of parent groups are also well known.

Examples of substituents are described in more detail below.

$C_{1-7}$  alkyl: The term " $C_{1-7}$  alkyl" as used herein, pertains to a monovalent moiety obtained by removing a hydrogen atom from a carbon atom of a hydrocarbon compound having from 1 to 7 carbon atoms, which may be aliphatic or alicyclic, and which may be 5 saturated or unsaturated (e.g. partially unsaturated, fully unsaturated). Thus, the term "alkyl" includes the sub-classes alkenyl, alkynyl, cycloalkyl, etc., discussed below.

Examples of saturated alkyl groups include, but are not limited 10 to, methyl ( $C_1$ ), ethyl ( $C_2$ ), propyl ( $C_3$ ), butyl ( $C_4$ ), pentyl ( $C_5$ ), hexyl ( $C_6$ ) and heptyl ( $C_7$ ).

Examples of saturated linear alkyl groups include, but are not limited to, methyl ( $C_1$ ), ethyl ( $C_2$ ), n-propyl ( $C_3$ ), n-butyl ( $C_4$ ), 15 n-pentyl (amyl) ( $C_5$ ), n-hexyl ( $C_6$ ) and n-heptyl ( $C_7$ ).

Examples of saturated branched alkyl groups include iso-propyl ( $C_3$ ), iso-butyl ( $C_4$ ), sec-butyl ( $C_4$ ), tert-butyl ( $C_4$ ), iso-pentyl ( $C_5$ ), and neo-pentyl ( $C_5$ ).

20  $C_{2-7}$  Alkenyl: The term " $C_{2-7}$  alkenyl" as used herein, pertains to an alkyl group having one or more carbon-carbon double bonds.

Examples of unsaturated alkenyl groups include, but are not 25 limited to, ethenyl (vinyl,  $-CH=CH_2$ ), 1-propenyl ( $-CH=CH-CH_3$ ), 2-propenyl (allyl,  $-CH-CH=CH_2$ ), isopropenyl (1-methylvinyl,  $-C(CH_3)=CH_2$ ), butenyl ( $C_4$ ), pentenyl ( $C_5$ ), and hexenyl ( $C_6$ ).

30  $C_{2-7}$  alkynyl: The term " $C_{2-12}$  alkynyl" as used herein, pertains to an alkyl group having one or more carbon-carbon triple bonds.

Examples of unsaturated alkynyl groups include, but are not limited to, ethynyl (ethinyl,  $-C\equiv CH$ ) and 2-propynyl (propargyl,  $-CH_2-C\equiv CH$ ).

$C_{3-7}$  cycloalkyl: The term " $C_{3-7}$  cycloalkyl" as used herein, pertains to an alkyl group which is also a cyclyl group; that is, a monovalent moiety obtained by removing a hydrogen atom from an alicyclic ring atom of a cyclic hydrocarbon (carbocyclic) compound, which moiety has from 3 to 7 carbon atoms, including from 3 to 7 ring atoms.

Examples of cycloalkyl groups include, but are not limited to, those derived from:

10       saturated monocyclic hydrocarbon compounds:  
cyclopropane ( $C_3$ ), cyclobutane ( $C_4$ ), cyclopentane ( $C_5$ ),  
cyclohexane ( $C_6$ ), cycloheptane ( $C_7$ ), methylcyclopropane ( $C_4$ ),  
dimethylcyclopropane ( $C_5$ ), methylcyclobutane ( $C_5$ ),  
dimethylcyclobutane ( $C_6$ ), methylcyclopentane ( $C_6$ ),  
15       dimethylcyclopentane ( $C_7$ ) and methylcyclohexane ( $C_7$ );  
          unsaturated monocyclic hydrocarbon compounds:  
cyclopropene ( $C_3$ ), cyclobutene ( $C_4$ ), cyclopentene ( $C_5$ ),  
cyclohexene ( $C_6$ ), methylcyclopropene ( $C_4$ ), dimethylcyclopropene  
( $C_5$ ), methylcyclobutene ( $C_5$ ), dimethylcyclobutene ( $C_6$ ),  
20       methylcyclopentene ( $C_6$ ), dimethylcyclopentene ( $C_7$ ) and  
methylcyclohexene ( $C_7$ ); and  
          saturated polycyclic hydrocarbon compounds:  
norcarane ( $C_7$ ), norpinane ( $C_7$ ), norbornane ( $C_7$ ).

25        $C_{3-20}$  heterocyclyl: The term " $C_{3-20}$  heterocyclyl" as used herein, pertains to a monovalent moiety obtained by removing a hydrogen atom from a ring atom of a heterocyclic compound, which moiety has from 3 to 20 ring atoms, of which from 1 to 10 are ring heteroatoms. Preferably, each ring has from 3 to 7 ring atoms,  
30       of which from 1 to 4 are ring heteroatoms.

In this context, the prefixes (e.g.  $C_{3-20}$ ,  $C_{3-7}$ ,  $C_{5-6}$ , etc.) denote the number of ring atoms, or range of number of ring atoms, whether carbon atoms or heteroatoms. For example, the term  
35       " $C_{5-6}$  heterocyclyl", as used herein, pertains to a heterocyclyl group having 5 or 6 ring atoms.

Examples of monocyclic heterocyclyl groups include, but are not limited to, those derived from:

N<sub>1</sub>: aziridine (C<sub>3</sub>), azetidine (C<sub>4</sub>), pyrrolidine

5 (tetrahydropyrrole) (C<sub>5</sub>), pyrrolidine (e.g., 3-pyrrolidine, 2,5-dihydropyrrole) (C<sub>5</sub>), 2H-pyrrole or 3H-pyrrole (isopyrrole, isoazole) (C<sub>5</sub>), piperidine (C<sub>6</sub>), dihydropyridine (C<sub>6</sub>), tetrahydropyridine (C<sub>6</sub>), azepine (C<sub>7</sub>);

O<sub>1</sub>: oxirane (C<sub>3</sub>), oxetane (C<sub>4</sub>), oxolane (tetrahydrofuran) (C<sub>5</sub>),

10 oxole (dihydrofuran) (C<sub>5</sub>), oxane (tetrahydropyran) (C<sub>6</sub>), dihydropyran (C<sub>6</sub>), pyran (C<sub>6</sub>), oxepin (C<sub>7</sub>);

S<sub>1</sub>: thiirane (C<sub>3</sub>), thietane (C<sub>4</sub>), thiolane (tetrahydrothiophene) (C<sub>5</sub>), thiane (tetrahydrothiopyran) (C<sub>6</sub>), thiepane (C<sub>7</sub>);

O<sub>2</sub>: dioxolane (C<sub>5</sub>), dioxane (C<sub>6</sub>), and dioxepane (C<sub>7</sub>);

15 O<sub>3</sub>: trioxane (C<sub>6</sub>);

N<sub>2</sub>: imidazolidine (C<sub>5</sub>), pyrazolidine (diazolidine) (C<sub>5</sub>), imidazoline (C<sub>5</sub>), pyrazoline (dihydropyrazole) (C<sub>5</sub>), piperazine (C<sub>6</sub>);

N<sub>1</sub>O<sub>1</sub>: tetrahydrooxazole (C<sub>5</sub>), dihydrooxazole (C<sub>5</sub>), tetrahydroisoxazole (C<sub>5</sub>), dihydroisoxazole (C<sub>5</sub>), morpholine (C<sub>6</sub>), tetrahydrooxazine (C<sub>6</sub>), dihydrooxazine (C<sub>6</sub>), oxazine (C<sub>6</sub>);

N<sub>1</sub>S<sub>1</sub>: thiazoline (C<sub>5</sub>), thiazolidine (C<sub>5</sub>), thiomorpholine (C<sub>6</sub>);

N<sub>2</sub>O<sub>1</sub>: oxadiazine (C<sub>6</sub>);

O<sub>1</sub>S<sub>1</sub>: oxathiole (C<sub>5</sub>) and oxathiane (thioxane) (C<sub>6</sub>); and,

25 N<sub>1</sub>O<sub>1</sub>S<sub>1</sub>: oxathiazine (C<sub>6</sub>).

Examples of substituted monocyclic heterocyclyl groups include those derived from saccharides, in cyclic form, for example, furanoses (C<sub>5</sub>), such as arabinofuranose, lyxofuranose, 30 ribofuranose, and xylofuranose, and pyranoses (C<sub>6</sub>), such as allopyranose, altropyranose, glucopyranose, mannopyranose, gulopyranose, idopyranose, galactopyranose, and talopyranose.

C<sub>5-20</sub> aryl: The term "C<sub>5-20</sub> aryl", as used herein, pertains to a

35 monovalent moiety obtained by removing a hydrogen atom from an aromatic ring atom of an aromatic compound, which moiety has

from 3 to 20 ring atoms. Preferably, each ring has from 5 to 7 ring atoms.

5 In this context, the prefixes (e.g. C<sub>3-20</sub>, C<sub>5-7</sub>, C<sub>5-6</sub>, etc.) denote the number of ring atoms, or range of number of ring atoms, whether carbon atoms or heteroatoms. For example, the term "C<sub>5-6</sub> aryl" as used herein, pertains to an aryl group having 5 or 6 ring atoms.

10 The ring atoms may be all carbon atoms, as in "carboaryl groups".

Examples of carboaryl groups include, but are not limited to, those derived from benzene (i.e. phenyl) (C<sub>6</sub>), naphthalene (C<sub>10</sub>), azulene (C<sub>10</sub>), anthracene (C<sub>14</sub>), phenanthrene (C<sub>14</sub>), naphthacene 15 (C<sub>18</sub>), and pyrene (C<sub>16</sub>).

20 Examples of aryl groups which comprise fused rings, at least one of which is an aromatic ring, include, but are not limited to, groups derived from indane (e.g. 2,3-dihydro-1H-indene) (C<sub>9</sub>), indene (C<sub>9</sub>), isoindene (C<sub>9</sub>), tetraline (1,2,3,4-tetrahydronaphthalene) (C<sub>10</sub>), acenaphthene (C<sub>12</sub>), fluorene (C<sub>13</sub>), phenalene (C<sub>13</sub>), acephenanthrene (C<sub>15</sub>), and aceanthrene (C<sub>16</sub>).

25 Alternatively, the ring atoms may include one or more heteroatoms, as in "heteroaryl groups". Examples of monocyclic heteroaryl groups include, but are not limited to, those derived from:

N<sub>1</sub>: pyrrole (azole) (C<sub>5</sub>), pyridine (azine) (C<sub>6</sub>);

30 O<sub>1</sub>: furan (oxole) (C<sub>5</sub>);

S<sub>1</sub>: thiophene (thiole) (C<sub>5</sub>);

N<sub>1</sub>O<sub>1</sub>: oxazole (C<sub>5</sub>), isoxazole (C<sub>5</sub>), isoxazine (C<sub>6</sub>);

N<sub>2</sub>O<sub>1</sub>: oxadiazole (furazan) (C<sub>5</sub>);

N<sub>3</sub>O<sub>1</sub>: oxatriazole (C<sub>5</sub>);

35 N<sub>1</sub>S<sub>1</sub>: thiazole (C<sub>5</sub>), isothiazole (C<sub>5</sub>);

$N_2$ : imidazole (1,3-diazole) ( $C_5$ ), pyrazole (1,2-diazole) ( $C_5$ ), pyridazine (1,2-diazine) ( $C_6$ ), pyrimidine (1,3-diazine) ( $C_6$ ) (e.g., cytosine, thymine, uracil), pyrazine (1,4-diazine) ( $C_6$ );

$N_3$ : triazole ( $C_5$ ), triazine ( $C_6$ ); and,

5  $N_4$ : tetrazole ( $C_5$ ).

Examples of heteroaryl which comprise fused rings, include, but are not limited to:

10  $C_9$  (with 2 fused rings) derived from benzofuran ( $O_1$ ), isobenzofuran ( $O_1$ ), indole ( $N_1$ ), isoindole ( $N_1$ ), indolizine ( $N_1$ ), indoline ( $N_1$ ), isoindoline ( $N_1$ ), purine ( $N_4$ ) (e.g., adenine, guanine), benzimidazole ( $N_2$ ), indazole ( $N_2$ ), benzoxazole ( $N_1O_1$ ), benzisoxazole ( $N_1O_1$ ), benzodioxole ( $O_2$ ), benzofurazan ( $N_2O_1$ ), benzotriazole ( $N_3$ ), benzothiofuran ( $S_1$ ), benzothiazole ( $N_1S_1$ ),

15 benzothiadiazole ( $N_2S$ );

16  $C_{10}$  (with 2 fused rings) derived from chromene ( $O_1$ ), isochromene ( $O_1$ ), chroman ( $O_1$ ), isochroman ( $O_1$ ), benzodioxan ( $O_2$ ), quinoline ( $N_1$ ), isoquinoline ( $N_1$ ), quinolizine ( $N_1$ ), benzoxazine ( $N_1O_1$ ), benzodiazine ( $N_2$ ), pyridopyridine ( $N_2$ ),

20 quinoxaline ( $N_2$ ), quinazoline ( $N_2$ ), cinnoline ( $N_2$ ), phthalazine ( $N_2$ ), naphthyridine ( $N_2$ ), pteridine ( $N_4$ );

21  $C_{11}$  (with 2 fused rings) derived from benzodiazepine ( $N_2$ );

22  $C_{13}$  (with 3 fused rings) derived from carbazole ( $N_1$ ), dibenzofuran ( $O_1$ ), dibenzothiophene ( $S_1$ ), carboline ( $N_2$ ),

25 perimidine ( $N_2$ ), pyridoindole ( $N_2$ ); and,

26  $C_{14}$  (with 3 fused rings) derived from acridine ( $N_1$ ), xanthene ( $O_1$ ), thioxanthene ( $S_1$ ), oxanthrene ( $O_2$ ), phenoxathiin ( $O_1S_1$ ), phenazine ( $N_2$ ), phenoxazine ( $N_1O_1$ ), phenothiazine ( $N_1S_1$ ), thianthrene ( $S_2$ ), phenanthridine ( $N_1$ ), phenanthroline ( $N_2$ ),

30 phenazine ( $N_2$ ).

31 The above groups, whether alone or part of another substituent, may themselves optionally be substituted with one or more groups selected from themselves and the additional substituents listed below.

Halo: -F, -Cl, -Br, and -I.

Hydroxy: -OH.

5 Ether: -OR, wherein R is an ether substituent, for example, a C<sub>1-7</sub> alkyl group (also referred to as a C<sub>1-7</sub> alkoxy group, discussed below), a C<sub>3-20</sub> heterocyclyl group (also referred to as a C<sub>3-20</sub> heterocyclyloxy group), or a C<sub>5-20</sub> aryl group (also referred to as a C<sub>5-20</sub> aryloxy group), preferably a C<sub>1-7</sub> alkyl group.

10

Alkoxy: -OR, wherein R is an alkyl group, for example, a C<sub>1-7</sub> alkyl group. Examples of C<sub>1-7</sub> alkoxy groups include, but are not limited to, -OMe (methoxy), -OEt (ethoxy), -O(nPr) (n-propoxy), -O(iPr) (isopropoxy), -O(nBu) (n-butoxy), -O(sBu) (sec-butoxy), 15 -O(iBu) (isobutoxy), and -O(tBu) (tert-butoxy).

20 Acetal: -CH(OR<sup>1</sup>)(OR<sup>2</sup>), wherein R<sup>1</sup> and R<sup>2</sup> are independently acetal substituents, for example, a C<sub>1-7</sub> alkyl group, a C<sub>3-20</sub> heterocyclyl group, or a C<sub>5-20</sub> aryl group, preferably a C<sub>1-7</sub> alkyl group, or, in the case of a "cyclic" acetal group, R<sup>1</sup> and R<sup>2</sup>, taken together with the two oxygen atoms to which they are attached, and the carbon atoms to which they are attached, form a heterocyclic ring having from 4 to 8 ring atoms. Examples of acetal groups include, but are not limited to, -CH(OMe)<sub>2</sub>, -CH(OEt)<sub>2</sub>, and 25 -CH(OMe)(OEt).

30 Hemiacetal: -CH(OH)(OR<sup>1</sup>), wherein R<sup>1</sup> is a hemiacetal substituent, for example, a C<sub>1-7</sub> alkyl group, a C<sub>3-20</sub> heterocyclyl group, or a C<sub>5-20</sub> aryl group, preferably a C<sub>1-7</sub> alkyl group. Examples of hemiacetal groups include, but are not limited to, -CH(OH)(OMe) and -CH(OH)(OEt).

35 Ketal: -CR(OR<sup>1</sup>)(OR<sup>2</sup>), where R<sup>1</sup> and R<sup>2</sup> are as defined for acetals, and R is a ketal substituent other than hydrogen, for example, a C<sub>1-7</sub> alkyl group, a C<sub>3-20</sub> heterocyclyl group, or a C<sub>5-20</sub> aryl group, preferably a C<sub>1-7</sub> alkyl group. Examples ketal groups include,

but are not limited to,  $-\text{C}(\text{Me})(\text{OMe})_2$ ,  $-\text{C}(\text{Me})(\text{OEt})_2$ ,  $-\text{C}(\text{Me})(\text{OMe})(\text{OEt})$ ,  $-\text{C}(\text{Et})(\text{OMe})_2$ ,  $-\text{C}(\text{Et})(\text{OEt})_2$ , and  $-\text{C}(\text{Et})(\text{OMe})(\text{OEt})$ .

5 Hemiketal:  $-\text{CR}(\text{OH})(\text{OR}^1)$ , where  $\text{R}^1$  is as defined for hemiacetals, and  $\text{R}$  is a hemiketal substituent other than hydrogen, for example, a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a  $\text{C}_{5-20}$  aryl group, preferably a  $\text{C}_{1-7}$  alkyl group. Examples of hemiacetal groups include, but are not limited to,  
10  $-\text{C}(\text{Me})(\text{OH})(\text{OMe})$ ,  $-\text{C}(\text{Et})(\text{OH})(\text{OMe})$ ,  $-\text{C}(\text{Me})(\text{OH})(\text{OEt})$ , and  $-\text{C}(\text{Et})(\text{OH})(\text{OEt})$ .

Oxo (keto, -one):  $=\text{O}$ .

15 Thione (thiocetone):  $=\text{S}$ .

Imino (imine):  $=\text{NR}$ , wherein  $\text{R}$  is an imino substituent, for example, hydrogen,  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a  $\text{C}_{5-20}$  aryl group, preferably hydrogen or a  $\text{C}_{1-7}$  alkyl group.

20 Examples of ester groups include, but are not limited to,  $=\text{NH}$ ,  $=\text{NMe}$ ,  $=\text{NET}$ , and  $=\text{NPh}$ .

Formyl (carbaldehyde, carboxaldehyde):  $-\text{C}(=\text{O})\text{H}$ .

25 Acyl (keto):  $-\text{C}(=\text{O})\text{R}$ , wherein  $\text{R}$  is an acyl substituent, for example, a  $\text{C}_{1-7}$  alkyl group (also referred to as  $\text{C}_{1-7}$  alkylacyl or  $\text{C}_{1-7}$  alkanoyl), a  $\text{C}_{3-20}$  heterocyclyl group (also referred to as  $\text{C}_{3-20}$  heterocyclacyl), or a  $\text{C}_{5-20}$  aryl group (also referred to as  $\text{C}_{5-20}$  arylacyl), preferably a  $\text{C}_{1-7}$  alkyl group. Examples of acyl groups include, but are not limited to,  $-\text{C}(=\text{O})\text{CH}_3$  (acetyl),  $-\text{C}(=\text{O})\text{CH}_2\text{CH}_3$  (propionyl),  $-\text{C}(=\text{O})\text{C}(\text{CH}_3)_3$  (t-butyryl), and  $-\text{C}(=\text{O})\text{Ph}$  (benzoyl, phenone).

Carboxy (carboxylic acid):  $-\text{C}(=\text{O})\text{OH}$ .

35

Thiocarboxy (thiocarboxylic acid):  $-\text{C}(=\text{S})\text{SH}$ .

Thiolocarboxy (thiolocarboxylic acid):  $-C(=O)SH$ .

Thionocarboxy (thionocarboxylic acid):  $-C(=S)OH$ .

5

Imidic acid:  $-C(=NH)OH$ .

Hydroxamic acid:  $-C(=NOH)OH$ .

10 Ester (carboxylate, carboxylic acid ester, oxycarbonyl):  
 $-C(=O)OR$ , wherein R is an ester substituent, for example, a  $C_{1-7}$  alkyl group, a  $C_{3-20}$  heterocyclyl group, or a  $C_{5-20}$  aryl group, preferably a  $C_{1-7}$  alkyl group. Examples of ester groups include, but are not limited to,  $-C(=O)OCH_3$ ,  $-C(=O)OCH_2CH_3$ ,  $-C(=O)OC(CH_3)_3$ ,  
15 and  $-C(=O)OPh$ .

Acyloxy (reverse ester):  $-OC(=O)R$ , wherein R is an acyloxy substituent, for example, a  $C_{1-7}$  alkyl group, a  $C_{3-20}$  heterocyclyl group, or a  $C_{5-20}$  aryl group, preferably a  $C_{1-7}$  alkyl group.

20 Examples of acyloxy groups include, but are not limited to,  $-OC(=O)CH_3$  (acetoxy),  $-OC(=O)CH_2CH_3$ ,  $-OC(=O)C(CH_3)_3$ ,  $-OC(=O)Ph$ , and  $-OC(=O)CH_2Ph$ .

25 Oxycarboyl oxy:  $-OC(=O)OR$ , wherein R is an ester substituent, for example, a  $C_{1-7}$  alkyl group, a  $C_{3-20}$  heterocyclyl group, or a  $C_{5-20}$  aryl group, preferably a  $C_{1-7}$  alkyl group. Examples of ester groups include, but are not limited to,  $-OC(=O)OCH_3$ ,  $-OC(=O)OCH_2CH_3$ ,  $-OC(=O)OC(CH_3)_3$ , and  $-OC(=O)OPh$ .

30 Amino:  $-NR^1R^2$ , wherein  $R^1$  and  $R^2$  are independently amino substituents, for example, hydrogen, a  $C_{1-7}$  alkyl group (also referred to as  $C_{1-7}$  alkylamino or di- $C_{1-7}$  alkylamino), a  $C_{3-20}$  heterocyclyl group, or a  $C_{5-20}$  aryl group, preferably H or a  $C_{1-7}$  alkyl group, or, in the case of a "cyclic" amino group,  $R^1$  and  
35  $R^2$ , taken together with the nitrogen atom to which they are attached, form a heterocyclic ring having from 4 to 8 ring

atoms. Amino groups may be primary ( $-\text{NH}_2$ ), secondary ( $-\text{NHR}^1$ ), or tertiary ( $-\text{NHR}^1\text{R}^2$ ), and in cationic form, may be quaternary ( $-\text{NR}^1\text{R}^2\text{R}^3$ ). Examples of amino groups include, but are not limited to,  $-\text{NH}_2$ ,  $-\text{NHCH}_3$ ,  $-\text{NHC}(\text{CH}_3)_2$ ,  $-\text{N}(\text{CH}_3)_2$ ,  $-\text{N}(\text{CH}_2\text{CH}_3)_2$ , and  $-\text{NHPh}$ .

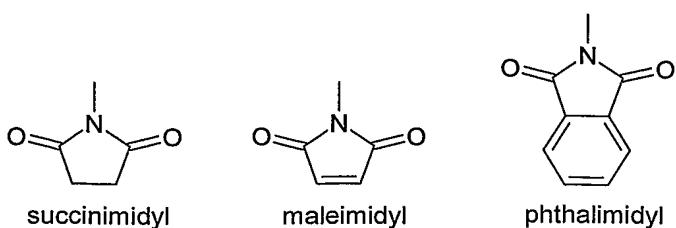
5 Examples of cyclic amino groups include, but are not limited to, aziridino, azetidino, pyrrolidino, piperidino, piperazino, morpholino, and thiomorpholino.

Amido (carbamoyl, carbamyl, aminocarbonyl, carboxamide):

10  $-\text{C}(=\text{O})\text{NR}^1\text{R}^2$ , wherein  $\text{R}^1$  and  $\text{R}^2$  are independently amino substituents, as defined for amino groups. Examples of amido groups include, but are not limited to,  $-\text{C}(=\text{O})\text{NH}_2$ ,  $-\text{C}(=\text{O})\text{NHCH}_3$ ,  $-\text{C}(=\text{O})\text{N}(\text{CH}_3)_2$ ,  $-\text{C}(=\text{O})\text{NHCH}_2\text{CH}_3$ , and  $-\text{C}(=\text{O})\text{N}(\text{CH}_2\text{CH}_3)_2$ , as well as amido groups in which  $\text{R}^1$  and  $\text{R}^2$ , together with the nitrogen atom 15 to which they are attached, form a heterocyclic structure as in, for example, piperidinocarbonyl, morpholinocarbonyl, thiomorpholinocarbonyl, and piperazinocarbonyl.

20 Thioamido (thiocarbamyl):  $-\text{C}(=\text{S})\text{NR}^1\text{R}^2$ , wherein  $\text{R}^1$  and  $\text{R}^2$  are independently amino substituents, as defined for amino groups. Examples of amido groups include, but are not limited to,  $-\text{C}(=\text{S})\text{NH}_2$ ,  $-\text{C}(=\text{S})\text{NHCH}_3$ ,  $-\text{C}(=\text{S})\text{N}(\text{CH}_3)_2$ , and  $-\text{C}(=\text{S})\text{NHCH}_2\text{CH}_3$ .

25 Acylamido (acylamino):  $-\text{NR}^1\text{C}(=\text{O})\text{R}^2$ , wherein  $\text{R}^1$  is an amide substituent, for example, hydrogen, a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a  $\text{C}_{5-20}$  aryl group, preferably hydrogen or a  $\text{C}_{1-7}$  alkyl group, and  $\text{R}^2$  is an acyl substituent, for example, a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a  $\text{C}_{5-20}$  aryl group, preferably hydrogen or a  $\text{C}_{1-7}$  alkyl group. Examples of acylamide groups include, but are not limited to,  $-\text{NHC}(=\text{O})\text{CH}_3$ ,  $-\text{NHC}(=\text{O})\text{CH}_2\text{CH}_3$ , and  $-\text{NHC}(=\text{O})\text{Ph}$ .  $\text{R}^1$  and  $\text{R}^2$  may together form a cyclic structure, as in, for example, succinimidyl, maleimidyl, and phthalimidyl:



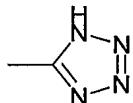
Aminocarbonyloxy:  $-\text{OC}(\text{=O})\text{NR}^1\text{R}^2$ , wherein  $\text{R}^1$  and  $\text{R}^2$  are independently amino substituents, as defined for amino groups.

5 Examples of aminocarbonyloxy groups include, but are not limited to,  $-\text{OC}(\text{=O})\text{NH}_2$ ,  $-\text{OC}(\text{=O})\text{NHMe}$ ,  $-\text{OC}(\text{=O})\text{NMe}_2$ , and  $-\text{OC}(\text{=O})\text{NET}_2$ .

Ureido:  $-\text{N}(\text{R}^1)\text{CONR}^2\text{R}^3$  wherein  $\text{R}^2$  and  $\text{R}^3$  are independently amino substituents, as defined for amino groups, and  $\text{R}^1$  is a ureido substituent, for example, hydrogen, a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a  $\text{C}_{5-20}$  aryl group, preferably hydrogen or a  $\text{C}_{1-7}$  alkyl group. Examples of ureido groups include, but are not limited to,  $-\text{NHCONH}_2$ ,  $-\text{NHCONHMe}$ ,  $-\text{NHCONHET}$ ,  $-\text{NHCONMe}_2$ ,  $-\text{NHCONET}_2$ ,  $-\text{NMeCONH}_2$ ,  $-\text{NMeCONHMe}$ ,  $-\text{NMeCONHET}$ ,  $-\text{NMeCONMe}_2$ , and  $-\text{NMeCONET}_2$ .

Guanidino:  $-\text{NH}-\text{C}(\text{=NH})\text{NH}_2$ .

Tetrazolyl: a five membered aromatic ring having four nitrogen atoms and one carbon atom,



Imino:  $=\text{NR}$ , wherein R is an imino substituent, for example, for example, hydrogen, a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a  $\text{C}_{5-20}$  aryl group, preferably H or a  $\text{C}_{1-7}$  alkyl group. Examples of imino groups include, but are not limited to,  $=\text{NH}$ ,  $=\text{NMe}$ , and  $=\text{NET}$ .

30 Amidine (amidino):  $-\text{C}(\text{=NR})\text{NR}_2$ , wherein each R is an amidine substituent, for example, hydrogen, a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$

heterocyclyl group, or a C<sub>5-20</sub> aryl group, preferably H or a C<sub>1-7</sub> alkyl group. Examples of amidine groups include, but are not limited to, -C(=NH)NH<sub>2</sub>, -C(=NH)NMe<sub>2</sub>, and -C(=NMe)NMe<sub>2</sub>.

5 Nitro: -NO<sub>2</sub>.

Nitroso: -NO.

Azido: -N<sub>3</sub>.

10 Cyano (nitrile, carbonitrile): -CN.

Isocyano: -NC.

15 Cyanato: -OCN.

Isocyanato: -NCO.

Thiocyanato (thiocyanato): -SCN.

20 Isothiocyanato (isothiocyanato): -NCS.

Sulphydryl (thiol, mercapto): -SH.

25 Thioether (sulfide): -SR, wherein R is a thioether substituent, for example, a C<sub>1-7</sub> alkyl group (also referred to as a C<sub>1-7</sub>alkylthio group), a C<sub>3-20</sub> heterocyclyl group, or a C<sub>5-20</sub> aryl group, preferably a C<sub>1-7</sub> alkyl group. Examples of C<sub>1-7</sub> alkylthio groups include, but are not limited to, -SCH<sub>3</sub> and -SCH<sub>2</sub>CH<sub>3</sub>.

30 Disulfide: -SS-R, wherein R is a disulfide substituent, for example, a C<sub>1-7</sub> alkyl group, a C<sub>3-20</sub> heterocyclyl group, or a C<sub>5-20</sub> aryl group, preferably a C<sub>1-7</sub> alkyl group (also referred to herein as C<sub>1-7</sub> alkyl disulfide). Examples of C<sub>1-7</sub> alkyl disulfide groups include, but are not limited to, -SSCH<sub>3</sub> and -SSCH<sub>2</sub>CH<sub>3</sub>.

Sulfine (sulfinyl, sulfoxide):  $-S(=O)R$ , wherein R is a sulfine substituent, for example, a  $C_{1-7}$  alkyl group, a  $C_{3-20}$  heterocyclyl group, or a  $C_{5-20}$  aryl group, preferably a  $C_{1-7}$  alkyl group.

Examples of sulfine groups include, but are not limited to,

5  $-S(=O)CH_3$  and  $-S(=O)CH_2CH_3$ .

Sulfone (sulfonyl):  $-S(=O)_2R$ , wherein R is a sulfone substituent, for example, a  $C_{1-7}$  alkyl group, a  $C_{3-20}$  heterocyclyl group, or a  $C_{5-20}$  aryl group, preferably a  $C_{1-7}$  alkyl group,

10 including, for example, a fluorinated or perfluorinated  $C_{1-7}$  alkyl group. Examples of sulfone groups include, but are not limited to,  $-S(=O)_2CH_3$  (methanesulfonyl, mesyl),  $-S(=O)_2CF_3$  (triflyl),  $-S(=O)_2CH_2CH_3$  (esyl),  $-S(=O)_2C_4F_9$  (nonaflyl),  $-S(=O)_2CH_2CF_3$  (tresyl),  $-S(=O)_2CH_2CH_2NH_2$  (tauryl),  $-S(=O)_2Ph$  15 (phenylsulfonyl, besyl), 4-methylphenylsulfonyl (tosyl), 4-chlorophenylsulfonyl (closyl), 4-bromophenylsulfonyl (brosyl), 4-nitrophenyl (nosyl), 2-naphthalenesulfonate (napsyl), and 5-dimethylamino-naphthalen-1-ylsulfonate (dansyl).

20 Sulfinic acid (sulfino):  $-S(=O)OH$ ,  $-SO_2H$ .

Sulfonic acid (sulfo):  $-S(=O)_2OH$ ,  $-SO_3H$ .

Sulfinate (sulfinic acid ester):  $-S(=O)OR$ ; wherein R is a sulfinate substituent, for example, a  $C_{1-7}$  alkyl group, a  $C_{3-20}$  heterocyclyl group, or a  $C_{5-20}$  aryl group, preferably a  $C_{1-7}$  alkyl group. Examples of sulfinate groups include, but are not limited to,  $-S(=O)OCH_3$  (methoxysulfinyl; methyl sulfinate) and  $-S(=O)OCH_2CH_3$  (ethoxysulfinyl; ethyl sulfinate).

30 Sulfonate (sulfonic acid ester):  $-S(=O)_2OR$ , wherein R is a sulfonate substituent, for example, a  $C_{1-7}$  alkyl group, a  $C_{3-20}$  heterocyclyl group, or a  $C_{5-20}$  aryl group, preferably a  $C_{1-7}$  alkyl group. Examples of sulfonate groups include, but are not limited to,  $-S(=O)_2OCH_3$  (methoxysulfonyl; methyl sulfonate) and  $-S(=O)_2OCH_2CH_3$  (ethoxysulfonyl; ethyl sulfonate).

Sulfinyloxy:  $-\text{OS}(\text{=O})\text{R}$ , wherein R is a sulfinyloxy substituent, for example, a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a  $\text{C}_{5-20}$  aryl group, preferably a  $\text{C}_{1-7}$  alkyl group. Examples of 5 sulfinyloxy groups include, but are not limited to,  $-\text{OS}(\text{=O})\text{CH}_3$  and  $-\text{OS}(\text{=O})\text{CH}_2\text{CH}_3$ .

Sulfonyloxy:  $-\text{OS}(\text{=O})_2\text{R}$ , wherein R is a sulfonyloxy substituent, for example, a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a 10  $\text{C}_{5-20}$  aryl group, preferably a  $\text{C}_{1-7}$  alkyl group. Examples of sulfonyloxy groups include, but are not limited to,  $-\text{OS}(\text{=O})_2\text{CH}_3$  (mesylate) and  $-\text{OS}(\text{=O})_2\text{CH}_2\text{CH}_3$  (esylate).

Sulfate:  $-\text{OS}(\text{=O})_2\text{OR}$ ; wherein R is a sulfate substituent, for 15 example, a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a  $\text{C}_{5-20}$  aryl group, preferably a  $\text{C}_{1-7}$  alkyl group. Examples of sulfate groups include, but are not limited to,  $-\text{OS}(\text{=O})_2\text{OCH}_3$  and  $-\text{SO}(\text{=O})_2\text{OCH}_2\text{CH}_3$ .

20 Sulfamyl (sulfamoyl; sulfinic acid amide; sulfonamide):  $-\text{S}(\text{=O})\text{NR}^1\text{R}^2$ , wherein  $\text{R}^1$  and  $\text{R}^2$  are independently amino substituents, as defined for amino groups. Examples of sulfamyl groups include, but are not limited to,  $-\text{S}(\text{=O})\text{NH}_2$ ,  $-\text{S}(\text{=O})\text{NH}(\text{CH}_3)$ ,  $-\text{S}(\text{=O})\text{N}(\text{CH}_3)_2$ ,  $-\text{S}(\text{=O})\text{NH}(\text{CH}_2\text{CH}_3)$ ,  $-\text{S}(\text{=O})\text{N}(\text{CH}_2\text{CH}_3)_2$ , and  $-\text{S}(\text{=O})\text{NHPh}$ .

25 Sulfonamido (sulfinamoyl; sulfonic acid amide; sulfonamide):  $-\text{S}(\text{=O})_2\text{NR}^1\text{R}^2$ , wherein  $\text{R}^1$  and  $\text{R}^2$  are independently amino substituents, as defined for amino groups. Examples of sulfonamido groups include, but are not limited to,  $-\text{S}(\text{=O})_2\text{NH}_2$ ,  $-\text{S}(\text{=O})_2\text{NH}(\text{CH}_3)$ ,  $-\text{S}(\text{=O})_2\text{N}(\text{CH}_3)_2$ ,  $-\text{S}(\text{=O})_2\text{NH}(\text{CH}_2\text{CH}_3)$ ,  $-\text{S}(\text{=O})_2\text{N}(\text{CH}_2\text{CH}_3)_2$ , and  $-\text{S}(\text{=O})_2\text{NHPh}$ .

30 Sulfamino:  $-\text{NR}^1\text{S}(\text{=O})_2\text{OH}$ , wherein  $\text{R}^1$  is an amino substituent, as defined for amino groups. Examples of sulfamino groups include, 35 but are not limited to,  $-\text{NHS}(\text{=O})_2\text{OH}$  and  $-\text{N}(\text{CH}_3)\text{S}(\text{=O})_2\text{OH}$ .

Sulfonamino:  $-\text{NR}^1\text{S}(\text{=O})_2\text{R}$ , wherein  $\text{R}^1$  is an amino substituent, as defined for amino groups, and  $\text{R}$  is a sulfonamino substituent, for example, a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a  $\text{C}_{5-20}$  aryl group, preferably a  $\text{C}_{1-7}$  alkyl group. Examples of 5 sulfonamino groups include, but are not limited to,  $-\text{NHS}(\text{=O})_2\text{CH}_3$  and  $-\text{N}(\text{CH}_3)\text{S}(\text{=O})_2\text{C}_6\text{H}_5$ .

Sulfinamino:  $-\text{NR}^1\text{S}(\text{=O})\text{R}$ , wherein  $\text{R}^1$  is an amino substituent, as defined for amino groups, and  $\text{R}$  is a sulfinamino substituent, 10 for example, a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a  $\text{C}_{5-20}$  aryl group, preferably a  $\text{C}_{1-7}$  alkyl group. Examples of sulfinamino groups include, but are not limited to,  $-\text{NHS}(\text{=O})\text{CH}_3$  and  $-\text{N}(\text{CH}_3)\text{S}(\text{=O})\text{C}_6\text{H}_5$ .

15 Phosphino (phosphine):  $-\text{PR}_2$ , wherein  $\text{R}$  is a phosphino substituent, for example,  $-\text{H}$ , a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a  $\text{C}_{5-20}$  aryl group, preferably  $-\text{H}$ , a  $\text{C}_{1-7}$  alkyl group, or a  $\text{C}_{5-20}$  aryl group. Examples of phosphino groups include, but are not limited to,  $-\text{PH}_2$ ,  $-\text{P}(\text{CH}_3)_2$ ,  $-\text{P}(\text{CH}_2\text{CH}_3)_2$ , 20  $-\text{P}(\text{t-Bu})_2$ , and  $-\text{P}(\text{Ph})_2$ .

Phospho:  $-\text{P}(\text{=O})_2$ .

25 Phosphinyl (phosphine oxide):  $-\text{P}(\text{=O})\text{R}_2$ , wherein  $\text{R}$  is a phosphinyl substituent, for example, a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a  $\text{C}_{5-20}$  aryl group, preferably a  $\text{C}_{1-7}$  alkyl group or a  $\text{C}_{5-20}$  aryl group. Examples of phosphinyl groups include, but are not limited to,  $-\text{P}(\text{=O})(\text{CH}_3)_2$ ,  $-\text{P}(\text{=O})(\text{CH}_2\text{CH}_3)_2$ , 30  $-\text{P}(\text{=O})(\text{t-Bu})_2$ , and  $-\text{P}(\text{=O})(\text{Ph})_2$ .

Phosphonic acid (phosphono):  $-\text{P}(\text{=O})(\text{OH})_2$ .

35 Phosphonate (phosphono ester):  $-\text{P}(\text{=O})(\text{OR})_2$ , where  $\text{R}$  is a phosphonate substituent, for example,  $-\text{H}$ , a  $\text{C}_{1-7}$  alkyl group, a  $\text{C}_{3-20}$  heterocyclyl group, or a  $\text{C}_{5-20}$  aryl group, preferably  $-\text{H}$ , a  $\text{C}_{1-7}$  alkyl group, or a  $\text{C}_{5-20}$  aryl group. Examples of phosphonate

groups include, but are not limited to,  $-P(=O)(OCH_3)_2$ ,  $-P(=O)(OCH_2CH_3)_2$ ,  $-P(=O)(O-t-Bu)_2$ , and  $-P(=O)(OPh)_2$ .

Phosphoric acid (phosphonooxy):  $-OP(=O)(OH)_2$ .

5

Phosphate (phosphonooxy ester):  $-OP(=O)(OR)_2$ , where R is a phosphate substituent, for example, -H, a  $C_{1-7}$  alkyl group, a  $C_{3-20}$  heterocyclyl group, or a  $C_{5-20}$  aryl group, preferably -H, a  $C_{1-7}$  alkyl group, or a  $C_{5-20}$  aryl group. Examples of phosphate groups include, but are not limited to,  $-OP(=O)(OCH_3)_2$ ,  $-OP(=O)(OCH_2CH_3)_2$ ,  $-OP(=O)(O-t-Bu)_2$ , and  $-OP(=O)(OPh)_2$ .

Phosphorous acid:  $-OP(OH)_2$ .

15 Phosphite:  $-OP(OR)_2$ , where R is a phosphite substituent, for example, -H, a  $C_{1-7}$  alkyl group, a  $C_{3-20}$  heterocyclyl group, or a  $C_{5-20}$  aryl group, preferably -H, a  $C_{1-7}$  alkyl group, or a  $C_{5-20}$  aryl group. Examples of phosphite groups include, but are not limited to,  $-OP(OCH_3)_2$ ,  $-OP(OCH_2CH_3)_2$ ,  $-OP(O-t-Bu)_2$ , and  $-OP(OPh)_2$ .

20 Phosphoramidite:  $-OP(OR^1)-NR^2_2$ , where  $R^1$  and  $R^2$  are phosphoramidite substituents, for example, -H, a (optionally substituted)  $C_{1-7}$  alkyl group, a  $C_{3-20}$  heterocyclyl group, or a  $C_{5-20}$  aryl group, preferably -H, a  $C_{1-7}$  alkyl group, or a  $C_{5-20}$  aryl group. Examples of phosphoramidite groups include, but are not limited to,  $-OP(OCH_2CH_3)-N(CH_3)_2$ ,  $-OP(OCH_2CH_3)-N(i-Pr)_2$ , and  $-OP(OCH_2CH_2CN)-N(i-Pr)_2$ .

25 30 Phosphoramidate:  $-OP(=O)(OR^1)-NR^2_2$ , where  $R^1$  and  $R^2$  are phosphoramidate substituents, for example, -H, a (optionally substituted)  $C_{1-7}$  alkyl group, a  $C_{3-20}$  heterocyclyl group, or a  $C_{5-20}$  aryl group, preferably -H, a  $C_{1-7}$  alkyl group, or a  $C_{5-20}$  aryl group. Examples of phosphoramidate groups include, but are not limited to,  $-OP(=O)(OCH_2CH_3)-N(CH_3)_2$ ,  $-OP(=O)(OCH_2CH_3)-N(i-Pr)_2$ , and  $-OP(=O)(OCH_2CH_2CN)-N(i-Pr)_2$ .

*Further substituent groups*

Particular substituent groups of interest are ion-chelating groups of formula  $[-(\text{CH}_2\text{CH}_2\text{O})_n\text{CH}_2\text{CH}_2\text{OCH}_3]$ ,  $[-\text{O}(\text{CH}_2\text{CH}_2\text{O})_n\text{OCH}_3]$ ,  $[-(\text{CH}_2\text{CH}(\text{R}^A)\text{O})_n\text{CH}_2\text{CH}_2\text{OCH}_3]$  and  $[-\text{O}(\text{CH}_2\text{CH}(\text{R}^A)\text{O})_n\text{OCH}_3]$ , wherein n is an integer from 0 to 10, preferably 2 to 10, more preferably 2 to 4, and  $\text{R}^A$  is  $\text{C}_{1-10}$  alkyl, preferably  $\text{C}_{1-2}$  alkyl, and wherein the ion chelating groups comprise side chains in ologomeric or polymeric structures.

10

The ion chelating side chains are based on the repeat unit  $[-\text{OCH}_2\text{CH}_2-]$ . Side chain branching and/or the inclusion of  $[-\text{OCH}_2\text{O}-]$  repeat-units, are advantageous to inhibit crystallisation after metal ion complexation. The side chains contain preferably 3 or more  $[-\text{OCH}_2\text{CH}_2-]$  and most preferably 3 units terminating in  $\text{OR}^A$  ( $\text{R}^A = \text{C}_{1-10}$  alkyl, e.g. methyl) containing 4 oxygen atoms for cation chelation. Crown ethers may also be designed accordingly. Other side chain designs may be made according to the specific need for cation binding.

15 20 Alternative design features could be incorporated into monomers and polymers to favour anion binding.

These substituent groups are discussed in detail in Reference 32.

25

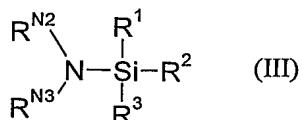
The  $\text{Ar}^1$ ,  $\text{Ar}^2$  and  $\text{Ar}^3$  groups as defined in Reference 32 are also of interest as  $\text{R}^{\text{N}1}$ ,  $\text{R}^{\text{N}2}$  and  $\text{R}^{\text{N}3}$  in the present invention.

*Compounds of formula II*

30 These compounds are either commercially available, or may be readily synthesised using known techniques.

## Compounds of formula III

Compounds of formula III:

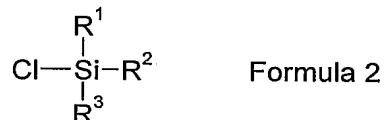


can be synthesised from compounds of Formula 1:



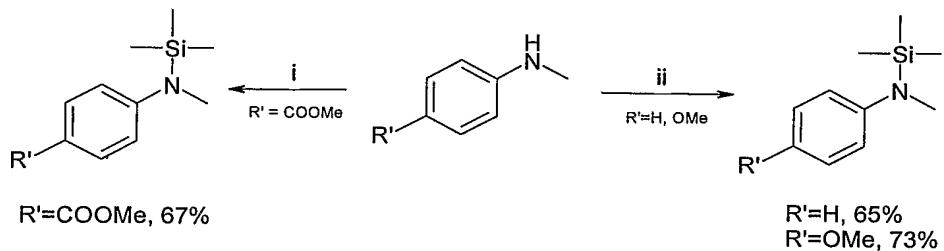
5

by methods known in the art. The method chosen will depend on the basicity of the amine of formula 1. Typically, the compound of formula 1 will be reacted with a base in organic solvent and then a compound of formula 2 added:



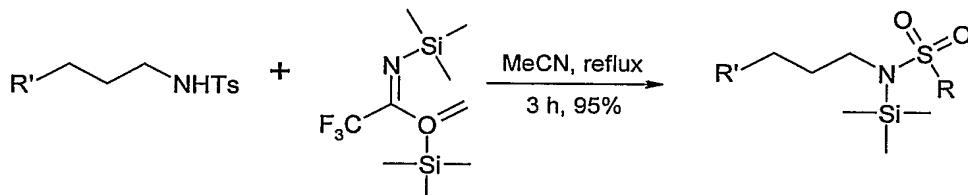
10

For example, some of the silylamines used in the examples below were prepared as follows from the free amine.

(i) TMSCl, NEt<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 17h

(ii) a. n-BuLi, THF, -78°C, 2 h. b. TMSCL, rt, 17h

15 The silylamines where R<sup>N2</sup> is sulfonyl were prepared as follows from a modified amine by heating with bis(trimethylsilyl)trifluoroacetamide (BSTFA).



The silylamines were purified by vacuum distillation. Once purified, their were handled under nitrogen at all times, and stored at -20°C.

5 If the desired compound of formula I is a tri-aryl amine, then the bi-aryl silyl amine of formula III, may itself be synthesised from a bi-aryl amine made by the method of the present invention.

10 *Further preferences*

The compounds of formula (I) may be oligomeric or polymeric in nature, as described in Reference 32. In particular, all of  $R^{N1}$ ,  $R^{N2}$  and  $R^{N3}$  may be substituted  $C_{5-20}$  aryl, preferably phenyl, with one of  $R^{N1}$ ,  $R^{N2}$  and  $R^{N3}$  being a side chain group, and the other 15 two of  $R^{N1}$ ,  $R^{N2}$  and  $R^{N3}$  being linked to form an oligomeric or polymeric backbone.

In some embodiments  $R^{N1}$  and  $R^{N2}$  are not linked by a single bond.

20  $R^{N1}$

$R^{N1}$  is, in some embodiments, preferably optionally substituted  $C_{5-7}$  aryl, more preferably optionally substituted phenyl.

$R^{N2}$

25  $R^{N2}$  is preferably selected from optionally substituted  $C_{5-20}$  aryl, optionally substituted  $C_{5-20}$  heterocyclyl, and optionally substituted sulfonyl. If  $R^{N2}$  is a sulfonyl group, then the sulfonyl substituent is preferably optionally substituted  $C_{1-7}$  alkyl.

30

$R^{N2}$  is more preferably selected from optionally substituted  $C_{5-20}$  aryl and optionally substituted  $C_{5-20}$  heterocyclyl, with optionally substituted  $C_{5-20}$  aryl (e.g. phenyl) being most preferred.

$R^{N3}$ 

$R^{N3}$  is preferably selected from optionally substituted  $C_{1-7}$  alkyl,  $C_{3-20}$  heterocyclyl and  $C_{5-20}$  aryl. If  $R^{N3}$  is selected from  $C_{1-7}$  alkyl, it is preferably  $C_{1-4}$  alkyl, and most preferably methyl.

5 If  $R^{N3}$  is selected from  $C_{5-20}$  aryl, it is preferably  $C_{5-7}$  aryl, and most preferably phenyl.

 $R^{N2}$  and  $R^{N3}$ 

When  $R^{N2}$  and  $R^{N3}$  together with the nitrogen atom to which they are attached form optionally substituted nitrogen-containing  $C_{3-20}$  heterocyclyl or  $C_{5-20}$  heteroaryl, they preferably form optionally substituted nitrogen-containing  $C_{5-20}$  heterocyclyl or heteroaryl (e.g. pyrrolyl, indolyl).

15  $R^1$ ,  $R^2$  and  $R^3$

$R^1$ ,  $R^2$  and  $R^3$  are preferably independently selected from optionally substituted  $C_{1-7}$  alkyl,  $C_{5-20}$  aryl,  $C_{3-20}$  heterocyclyl and  $C_{1-7}$  alkoxy, or two of  $R^1$ ,  $R^2$  and  $R^3$ , together with the silicon atom to which they are attached, may form a silicon containing  $C_{5-7}$  heterocyclyl group. It is more preferred that  $R^1$ ,  $R^2$  and  $R^3$  are independently selected from optionally substituted  $C_{1-7}$  alkyl,  $C_{5-20}$  aryl and  $C_{3-20}$  heterocyclyl, with optionally substituted  $C_{1-7}$  alkyl being most preferred.

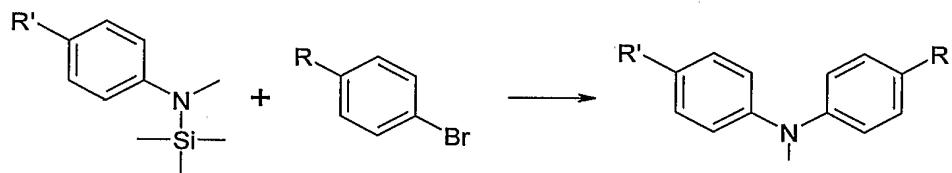
25 Examples of preferred  $\text{SiR}^1\text{R}^2\text{R}^3$  groups include TMS, TES, TIPS, TDDMS, TBDPS and 1-methylsilacyclobutane.

*Optional substituents*

The optional substituents for  $R^{N1}$ ,  $R^{N2}$  and  $R^{N3}$  when they are  $C_{5-20}$  aryl groups, for example phenyl, include, but are not limited to,  $C_{1-7}$  alkyl,  $C_{1-7}$  alkoxy and  $C_{1-7}$  alkyl ester, of which, in some embodiments,  $C_{1-7}$  alkoxy (e.g. OMe) and  $C_{1-7}$  alkyl ester (e.g. COOMe) are preferred.

Examples**General Method**

Flame dried cesium carbonate (228 mg, 0.7 mmol, 1.4 eq), aryl bromide (0.5 mmol), palladium acetate (2.8 mg, 0.012 mmol, 2.5 mol%) and di-*tert*-butyl biphenylphosphine (7.5 mg, 0.025 mmol, 5 mol%) were placed in a 10 cm<sup>3</sup> stainless steel cell and the cell sealed. The cell was evacuated and refilled with nitrogen (three cycles). The silylamine (1.2 eq) was injected through the inlet port and the cell connected to the CO<sub>2</sub> line and charged 10 with CO<sub>2</sub> (99.9995% - further purified over an Oxisorb<sup>RTM</sup> catalyst) to approximately 760 psi (volume ca. 1 cm<sup>3</sup> liquid carbon dioxide). The cell was heated to 100°C and the pressure adjusted to the desired pressure by the addition of further CO<sub>2</sub>. The reagents were maintained at this temperature and pressure 15 for the desired time before the cell was allowed to cool to room temperature. The contents of the cell were vented into ethyl acetate (50 cm<sup>3</sup>), and once atmospheric pressure had been reached, the cell was opened and washed with further ethyl acetate (3 x 10 cm<sup>3</sup>). The combined organic fractions were 20 filtered and concentrated *in vacuo* to furnish the crude material that was purified by flash column chromatography.

**Example 1**

25 The reaction was carried out as described in the general method.

- (a) R=COOMe, R'=COOMe, 3000 psi, 17 hours: Yield 84%
- (b) R=COOMe, R'=COOMe, 1800 psi, 17 hours: Yield 69%
- (c) R=COOMe, R'=OMe, 3000 psi, 17 hours: Yield 40%
- 30 (d) R=COOMe, R'=OMe, 1800 psi, 48 hours: Yield 77%
- (e) R=COOMe, R'=H, 3000 psi, 17 hours: Yield 28%
- (f) R=COOMe, R'=H, 1800 psi, 48 hours: Yield 76%
- (g) R=H, R'=COOMe, 1800 psi, 17 hours: Yield 77%

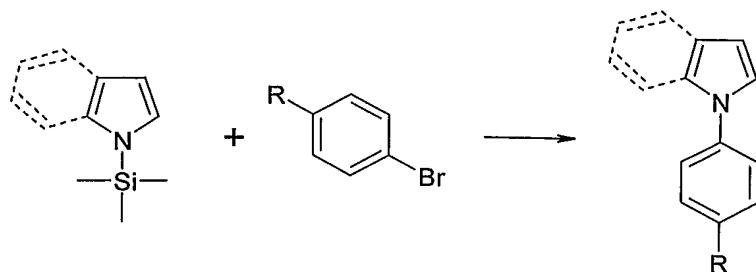
- (h) R=H, R'=H, 1800 psi, 48 hours: Yield 55%
- (i) R=H, R'=OMe, 1800 psi, 48 hours: Yield 66%
- (j) R=OMe, R'=COOMe, 1800 psi, 17 hours: Yield 57%
- (k) R=OMe, R'=H, 1800 psi, 48 hours: Yield 25%
- 5 (l) R=OMe, R'=OMe, 1800 psi, 48 hours: Yield 25%

As a comparison, the reaction was carried out with the same reagents in toluene, as follows. To an oven dried Schlenk tube under nitrogen was added cesium carbonate (228 mg, 0.7 mmol, 1.4 eq) and the cesium carbonate was flame dried under vacuum with stirring. Methyl bromobenzoate (108 mg, 0.5 mmol), palladium acetate (5.6 mg, 0.024 mmol, 5 mol%) and di-*tert*-butyl biphenylphosphine (15 mg, 0.05 mmol, 10 mol%) were added and the Schlenk tube sealed, and evacuated and refilled with nitrogen (3 cycles). A solution of the silylamine (1.2 eq) in dry toluene (1.5 cm<sup>3</sup>) was added and the reaction mixture heated at 100°C for the desired time. The reaction mixture was allowed to cool to room temperature. The mixture was filtered and concentrated *in vacuo* to furnish the crude material which was purified by flash column chromatography. The yields are shown in Table 1, with the time for each experiment in parentheses.

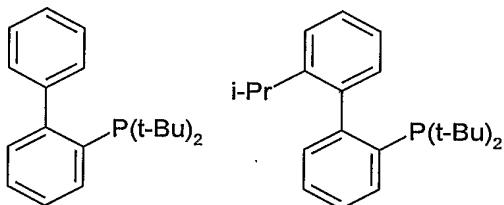
Table 1

	R=COOMe	R=H	R=OMe
R'=COOMe	66 (34h)	41 (17h)	63 (17h)
R'=H	65 (54h)	12 (17h)	8 (17h)
R'=OMe	72 (54h)	25 (17h)	7 (17h)

## Example 2



The reaction was carried out as described in the general method, with the R group and either the N-trimethylsilyl-pyrrole or 5 indole as shown in Table 3, with the yields expressed in %. The reactions were carried out at ca. 1800 psi for 17 hours. The catalyst ligand used was either:



A

B

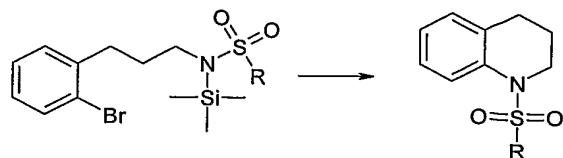
wherein ligand A is that described in the general method.

10

Table 2

Substrate	X	Yield (%)	
		A	B
Pyrrole	COOMe	59	75
	H	11	46
	OMe	7	30
Indole	COOMe	70	88
	H	68	70
	OMe	25	50

**Example 3**



The reaction is carried out as described in the general method,  
5 wherein the starting material is added at the silylamine stage. An additive (1.2 eq) was sometimes added (see table 3) at the same time as the  $\text{Cs}_2\text{CO}_3$ . The reaction was carried out at 1800 psi for the length of time as shown in Table 3.

Table 3

R	Additive	Time (hours)	Yield (%)
	-	17	43
	-	41	61
	KF	41	57
-CH <sub>3</sub>	-	17	55
-CH <sub>3</sub>	-	41	28
-CH <sub>3</sub>	KF	17	72

10

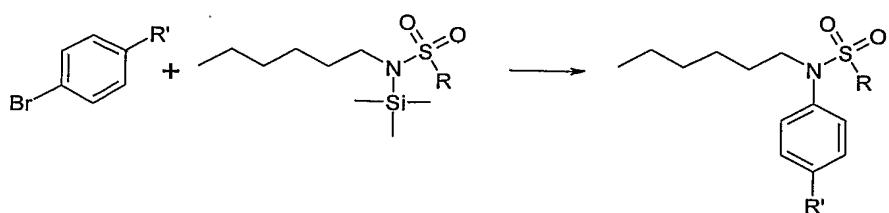
As a comparison, the reaction was also carried out where the starting material did not bear the trimethyl silyl group, as shown in Table 4:

Table 4

R	Additive	Time (hours)	Yield (%)
	-	17	20
-CH <sub>3</sub>	-	17	22

15

## Example 4



5 The reaction is carried out as described in the general method, and an additive (1.2 eq) was sometimes added (see table 5) at the same time as the  $\text{Cs}_2\text{CO}_3$ . The reaction was carried out at 1800 psi for the length of time as shown in Table 5.

Table 5

R	R'	Additive	Time (hours)	Yield (%)
$-\text{CH}_3$	COOMe	-	17	15
$-\text{CH}_3$	COOMe	KF	17	56
$-\text{CH}_3$	COOMe	KF	41	55

**References**

(all of which are herein incorporates by reference)

(1) M. Stolka, J. F. Yanus, and D. M. Pai, *J. Phys. Chem.*, 1984, **88**, 4707-4714.

5 (2) E. Ueta, H. Nakano, and Y. Shirota, *Chem. Lett.*, 1994, 2397.

(3) Y. Kuwabara, H. Ogawa, H. Inada, N. Noma, and Y. Shirota, *Adv. Mater.*, 1994, **6**, 677.

(4) M. Strukelj, R. H. Jordan, and A. Dodabalapur, *J. Am. Chem. Soc.*, 1996, **118**, 1213-1214.

10 (5) A. Kitani, M. Kaya, J. Yano, K. Yoshikawa, and K. Sasaki, *Synth. Met.*, 1987, **18**, 341-346.

(6) F.-L. Lu, F. Wudl, M. Nowak, and A. J. Heeger, *J. Am. Chem. Soc.*, 1986, **108**, 8311-13.

(7) A. G. MacDiarmid, J. C. Chiang, A. F. Richter, and A. J. Epstein, *Synth. Met.*, 1987, **18**, 285-290.

15 (8) A. G. MacDiarmid, and A. J. Epstein, *Faraday Discuss. Chem. Soc.*, 1989, **88**, 317-332.

(9) A. G. MacDiarmid, and A. J. Epstein, *Science and Applications of Conducting Polymers*; Hilger: New York, 1991.

20 (10) A. Ray, A. F. Richter, D. L. Kershner, and A. J. Epstein, *Synth. Met.*, 1989, **29**, 141-150.

(11) D. Vachon, R. O. Angus, Jr., F.-L. Lu, M. Nowak, Z. X. Liu, H. Schaffer, F. Wudl, and A.J. Heeger, *Synth. Met.*, 1987, **18**, 297-302.

25 (12) R. S. Oakes, A. A. Clifford, and C. M. Rayner, *J. Chem. Soc. Perkin Trans 1*, 2001, 917-941.

(13) P. G. Jessop, and W. Leitner *Chemical Synthesis Using Supercritical Fluids*; Wiley-VCH: Weinheim, 1999.

(14) A. I. Cooper, *Adv. Mater.*, 2001, **13**, 1111-1114.

30 (15) M. A. Carroll, and A. B. Holmes, *Chem. Commun.*, 1998, 1395-1396.

(16) T. R. Early, R. S. Gordon, M. A. Carroll, A. B. Holmes, R. E. Shute, and I. F. McConvey, *Chem. Commun.*, 2001, 1966-1967.

(17) R. S. Gordon, and A. B. Holmes, *Chem. Commun.*, 2002, 640-641.

(18) S. V. Ley, C. Ramarao, R. S. Gordon, A. B. Holmes, A. J. Morrison, I. F. McConvey, I. M. Shirley, S. C. Smith, and M. D. Smith, *Chem. Commun.*, 2002, 1134-1135.

35 (19) N. Sundararajan, S. Yang, K. Ogino, S. Valiyaveettil, J. G. Wang, X. Y. Zhou, C. K. Ober, S. K. Obendorf, and R. D. Allen, *Chem. Mater.*, 2000, **12**, 41-48.

(20) Y. C. Bae, K. Douki, T. Y. Yu, J. Y. Dai, D. Schmaljohann, H. Koerner, C. K. Ober, and W. Conley, *Chem. Mater.*, 2002, **14**, 1306-1313.

5 (21) J. M. D. E. Hoggan, R. G. Carbonell, *Polym. Prepr. Am. Chem. Soc. Div. PMSE, Part 2 Aug 22, 1999*, 218.

(22) S. L. Wells, and J. DeSimone, *Angew. Chem. Int. Ed. Engl.*, 2001, **40**, 518-527.

10 (23) F. Gaspar, T. Lu, R. Santos, B. Al-Duri, A. B. Holmes, G. Leeke, W. T. S. Huck, C. K. Luscombe, and J. Seville, Patterned deposition using compressed carbon dioxide, 2003, EP 1 341 616.

(24) J. Lindley, *Tetrahedron*, 1984, **40**, 1433-1456.

15 (25) H. L. Aalten, G. van Koten, and D. M. Grove, *Tetrahedron*, 1989, **45**, 5565-5578.

(26) A. J. Paine, *J. Am. Chem. Soc.*, 1987, **109**, 1496-1502.

(27) H. Weingarten, *J. Org. Chem.*, 1964, **29**, 975-977.

20 (28) S. L. Buchwald, and A. S. Guram, Preparation of arylamines, 1994, US 5 576 460.

(29) J. P. Wolfe, S. Wagaw, J.-F. Marcoux, and S. L. Buchwald, *Acc. Chem. Res.*, 1998, **31**, 805-818.

(30) J. F. Hartwig, *Angew. Chem. Int. Ed. Engl.*, 1998, **37**, 2046-2047.

25 (31) B. Yang, and S. L. Buchwald, *J. Organometallic Chem.*, 1999, **576**, 125-146; A. R. Muci and S. L. Buchwald in *Topics in Current Chemistry: Cross Coupling Reactions*, Vol. 219, Springer-Verlag, Berlin, 2002.

(32) A. B. Holmes, and T. Park, Electroactive polyarylamine-type compositions, 2002, WO 02/051958.

30 (33) C. Salvatore, Light emissive polymer blends and light emissive devices made from the same, 2003, EP 1 326 942.

(34) J. J. M. Halls, C. A. Walsh, N. C. Greenham, E. A. Marseglia, R. H. Friend, S. C. Moratti, and A. B. Holmes, *Nature*, 1995, **376**, 498-500.

(35) A. Fürstner, L. Ackermann, K. Beck, H. Hori, D. Kock, K. Langermann, M. Liebl, C. Six, and W. Leitner, *J. Am. Chem. Soc.*, 2001, **123**, 9000-9006.

35 (36) A. B. Holmes, R. S. Gordon, and T. R. Early, WO 03/009936.

(37) A. Baiker, *Chem. Rev.*, 1999, **99**, 453-474 (p. 455)

(38) Shezad, N., Oakes, R. S., Clifford, A. A., and Rayner, C. M., *Chemical Industries (Dekker)* 2001, 82(Catalysis of Organic Reactions), 459-464

(39) N. Shezad, A.A. Clifford, and C.M. Rayner, *Green Chemistry* 2002, 4 (1), 64-67

(40) WO96/01304

(41) WO95/22591

5 (42) WO94/20444

(43) WO94/06738

(44) EP 0 652 202

(45) US 6,156,933

(46) Nolan, *Ionic liquids as green solvents: progress and prospects*,  
10 ACS Symposium Series, 2003, 856, 323-341

(47) A. Deagostino, C. Prandi and P. Venurello, *Org. Lett.*, 2003, 5, 3815-3817

(48) *New Aspects in Phosphorus Chemistry II*, *Top. Curr. Chem.*, 2003, 223, 1-44